

STUDY MODULE DESCRIPTION FORM		
Name of the module/subject Chemistry		Code 1010601121010700060
Field of study Mechanical Engineering	Profile of study (general academic, practical) (brak)	Year /Semester 1 / 2
Elective path/specialty -	Subject offered in: Polish	Course (compulsory, elective) obligatory
Cycle of study: First-cycle studies	Form of study (full-time, part-time) full-time	
No. of hours Lecture: 1 Classes: - Laboratory: - Project/seminars: -		No. of credits 1
Status of the course in the study program (Basic, major, other) (brak)		(university-wide, from another field) (brak)
Education areas and fields of science and art the sciences		ECTS distribution (number and %) 1 100%
Responsible for subject / lecturer: Dr Maciej Galiński email: maciej.galinski@put.poznan.pl tel. 61 665-2310 Faculty of Chemical Technology ul. Piotrowo 3A, 60-965 Poznań		
Prerequisites in terms of knowledge, skills and social competencies:		
1	Knowledge	Basic information on the structure of matter, physical and chemical processes concerning. Basic information on chemistry.
2	Skills	Description on observed phenomena, drawing the conclusions, analysing of the results.
3	Social competencies	Ability of the self-learning, analysing and verification.
Assumptions and objectives of the course: Reminder basic chemical phenomena, structure of matter. Introduction to electrochemistry, Description of the corrosion processes its mechanism and prevention. Energy storage devices.		
Study outcomes and reference to the educational results for a field of study		
Knowledge: 1. Has a basic knowledge in chemistry, in the construction of the periodic table and properties of the elements, the theory of chemical bonding, organic and inorganic compounds, types of chemical reactions, chemical analysis: - [K1A_W03] 2. in the extent necessary for an understanding of lectures on metallic and non-metallic materials, environmental protection, fuels and lubricants, building materials and soil, biomechanics and biological materials processed by agricultural and food machinery - [K1A_W03]		
Skills: 1. Is able to obtain information from the literature, internet, databases and other sources. Can integrate the information to interpret and learn from them, create and justify opinions. - [K1A_U03] 2. Has the ability to self-educate using modern teaching tools such as remote lectures, webpages and databases, educational software, electronic books. - [K1A_U06]		
Social competencies: 1. Understands the need and knows the possibilities of lifelong learning - [K1A_K01]		
Assessment methods of study outcomes		
Test		
Course description		

Atomic nucleus composition. Nucleons, nuclides, chemical element, isotope, mol, position in the periodic table, mass number, atomic number, subatomic particles. Natural radioactive decays. Ionizing radiation - ? properties of ?????and???radiations. Detection of irradiation - Geigera-Mullera counter. Scintillating Counter ? construction and operations.

Types of solution concentrations. Electrolytes. Electrolytic dissociation. Law of Mass Action. Equilibrium constant. Solubility product. Poorly soluble compounds. Conductivity of the electrolytes ? comparison with metals.

Temperature dependencies of the conductivities of the electrolytes. Water hardness ? permanent and temporary Limescale. Methods of removing water hardness.

Reduction and Oxidation. The concept of half-cell (electrode) in electrochemistry. Primary and secondary cells, standard electrode potentials, - Nernst equation. Current flow through the electrode Potential of the deposition, overpotential.

Types and construction of the galvanic cells.

Corrosion, Types of corrosion Methods of protecting. Types of the protecting layers. Chemical energy sources. Construction and types of the primary and rechargeable batteries. Description of examples of typical rechargeable batteries. Fuels cells, Supercapacitors, Recycling of energy storage devices principles.

Commercial methods of metal production: sodium, potassium, zinc, aluminum, copper

Electrochemical method of metal refinement.

Basic bibliography:

1. 1. A. Bielański, Podstawy chemii nieorganicznej, Wydanie 6, PWN. 2012
2. 2. K. Pazdro. Podręcznik do kształcenia rozszerzonego t 1-4. Oficyna Edukacyjna

Additional bibliography:

1. 1. K. Pigoń, Z. Ruziewicz, Chemia Fizyczna, PWN Warszawa 2005
2. 2. P. Atkins, Chemia Fizyczna, PWN, Warszawa 2001

Result of average student's workload

Activity	Time (working hours)	
1. Participation in lectures	15	
2. Preparation to test	7	
3. Consultaion	1	
4. Participation in test	1	
Student's workload		
Source of workload	hours	ECTS
Total workload	24	1
Contact hours	17	0
Practical activities	0	0